Soft Matter Complex System

A single component system:

The structures of soft matter are determined by their free volume that is restricted by the geometries.

Colloid : spherical excluded volume \rightarrow close-packed crystal Liquid crystal : rod-like excluded volume \rightarrow nematic interaction Polymer : segment excluded volume \rightarrow self-avoiding chain Membrane : 2D sheet \rightarrow membrane fluctuations

Mixture of soft matter :

Competition of the free volume brings phase transition.

Depletion interaction

colloid and polymer mixture colloid and liquid crystal mixture

Depletion Interaction



Depletion zone surrounding spherical colloid particle

Competition between mixing entropy S_m and translational entropy S_t

$$S_m > S_t$$
 Miscible $S_m < S_t$ Aggregation

Thermodynamics of spherical colloid particles + polymer system



semipermeable membranes

H.N.W. Lekkerkerker et al., Europhys. Lett. 20, 559 (1992)

Ground Canonical Potential of the system III: Ω

$$\Omega = F - N_p \mu_p$$

F: Helmholtz Free Energy of System N_p : Number of Polymer Chains μ_p : Chemical Potential of Polymer

Grand Potential has the standard relationship

$$\Omega = -k_B T \ln \Xi$$
$$\Xi(N_c, \mu_p, V) = \sum_{N_p=0}^{\infty} z_p^{N_p} Z(N_c, N_p, V)$$

 $Z(N_c, N_p, V)$: Canonical partition function z_p : Activity of polymer chains

$$z_p = \exp(\mu_p / k_B T)$$

In this system, number of polymer chain, N_p , is variable, but determined by the chemical potential μp . Then we can deal with the system as canonical ensemble of colloids.

$$\Xi(N_c,\mu_p,V) = \frac{1}{N_c!} \int d\boldsymbol{r}^{N_c} \exp[-W(\boldsymbol{r}^{N_c};\mu_p)/k_BT]$$

 $W(\mathbf{r}^{N_c}; \mu_p)$ is the interaction potential of colloids depending on the configuration of colloids and chemical potential of polymers.

$$W = U_c - \Pi_p(\mu_p) V_{free}(\mathbf{r}^{N_c})$$

 $U_{\rm c}$: Inter-colloids potential without polymer $\Pi_{\rm p}$: Osmotic pressure of the reservoir (polymer solution) $V_{\rm free}$: Free volume of polymer chains

$$V_{free} = \alpha V$$

 $\alpha(\phi_{\rm c}, \delta/a)$: Free volume fraction

$$\Xi(N_c, \mu_p, V) = \frac{1}{N_c!} \int d\mathbf{r}^{N_c} \exp[-(U_c - \Pi_p \alpha V) / k_B T]$$
$$\sim \frac{1}{N_c!} \exp[-(U_c - \Pi_p \alpha V) / k_B T]$$

$$\Omega = U_c - k_B T (N_c \ln N_c - N_c) - \Pi_p \alpha V = F_c (N_c, V) - \Pi_p \alpha V$$

 $F_{\rm c}(N_{\rm c},V)$: Helmholtz free energy for pure colloidal system

Helmholtz free energy of the system is given by

$$\begin{split} F &= \Omega + N_p \mu_p \\ &= F_c(N_c,V) + N_p \mu_p - \Pi_p \alpha V \\ &= F_c(N_c,V) + F_p(N_p,\alpha V) \end{split}$$

Helmhortz Free energy of pure polymer solution

$$F_p(N_p, \alpha V) \cong k_B T n V \ln \frac{n}{\alpha}$$
$$n = N_p / V$$

Helmhortz Free energy of pure colloidal suspension

$$F_c(N_c, V) = -\int \Pi_c dV$$

 $\Pi_{\rm c}$ is the osmotic pressure of pure colloidal solution.

Using hard-sphere compressibility $\xi = \frac{\Pi_c V}{N_c k_B T}$

$$F_c(N_c,V) = \frac{3k_BT}{4\pi a^3}V\phi_c\int\frac{\zeta}{\phi_c}d\phi_c$$

For fluid phase

$$\zeta(\text{fluid}) = \frac{1 + \phi_c + \phi_c^2 - \phi_c^3}{(1 - \phi_c)^3}$$

For crystal phase (FCC)

$$\xi(\text{solid}) = \frac{1 + \phi_c + \phi_c^2 - 0.67825\phi_c^3 - \phi_c^4 - 0.5\phi_c^5 - 6.028e^{\zeta(7.9 - 3.9\zeta)}\phi_c^6}{1 - 3\phi_c + 3\phi_c^2 - 1.04305\phi_c^3}$$

In order to determine the coexistence of phase 1 and phase 2, we have to solve the following equations,

$$\mu_C(\phi_c^1, n_R) = \mu_C(\phi_c^2, n_R)$$
$$\Pi(\phi_c^1, n_R) = \Pi(\phi_c^2, n_R)$$

volume fraction of colloids : $\phi_c^1 \quad \phi_c^2$ number density of polymer chain : $n_R = n/\alpha = N_p / \alpha V$

In order to calculate the phase diagram, we must obtain the chemical potentials of the colloid and polymer (μ_c, μ_p) $\mu_i = (\partial \Omega / \partial N_i)_{V, N_{j(j\neq i)}}$, and the total osmotic pressure $\Pi = -(\partial \Omega / \partial V)_{N_c, \mu_p}$.

$$(\mu_{c} - \mu_{c}^{0})/k_{B}T = \int \frac{Z}{\phi_{c}} d\phi_{c} + Z - \frac{4}{3}\pi a^{3} \frac{n}{\alpha} \frac{d\alpha}{d\phi_{c}}$$
$$(\mu_{p} - \mu_{p}^{0})/k_{B}T = \ln(\frac{n}{\alpha})$$
$$\frac{4\pi a^{3}\Pi}{3k_{B}T} = \phi_{c}Z + \frac{4}{3}\pi a^{3} \frac{n}{\alpha}(\alpha - \phi_{c}\frac{d\alpha}{d\phi})$$

Next, we need the free volume fraction $\alpha(\phi)$.

$$\alpha = (1 - \phi_c) \exp[-A\gamma - B\gamma^2 - C\gamma^3]$$

$$\gamma = \phi_c / (1 - \phi_c) \quad A = 3\xi + 3\xi^2 + \xi^3 \quad B = \frac{9}{2}\xi^2 + 3\xi^3 \quad C = 3\xi^3 \quad \xi = \delta / a$$

Here we introduce the effective polymer number density $n_R = n / \alpha = N_p / \alpha V$,

$$n_R = \exp[(\mu_p - \mu_p^0)/k_B T]$$

Then we obtain expressions for μ_c and Π which are functions of ϕ and n_R . In order to calculate the phase diagram, we only need to solve

$$\mu_{C}(\phi_{c}^{1}, n_{R}) = \mu_{C}(\phi_{c}^{2}, n_{R}) \qquad \Pi(\phi_{c}^{1}, n_{R}) = \Pi(\phi_{c}^{2}, n_{R})$$

at given $n_{\rm R}$. The polymer concentrations in these phases are then found from

$$n_1 = \alpha(\phi_c^1) n_R \qquad n_2 = \alpha(\phi_c^2) n_R$$



Comparison between Theoretical Prediction and Experiments

S.M. Ilett, et al., Phys. Rev. E 51. 1344 (1995).

Phase behavior of PMMA colloidal particles (220 nm) + PS (R_g =54 nm) system





liquid (middle) gas (top)

crystal (bottom) gas (top)

Depletion Interaction Probed with a Line Optical Tweezers

The laser focus is scanned back and forth along a line in the focal plane of the objective at 180 Hz. Under the such optical field, the colloidal particles freely diffuse along the line while is strongly confined in other two directions.



The effective interaction potential between the two particles is related to the probability, P(r), of finding the two spheres with centers separated by r through the Boltzmann relation

 $P(r) \propto \exp(-U(r)/k_B T)$

R. Verma, et al., Macromolecules 33, 177 (2000).

Optical tweezers



Individual rays of light emitted from the laser will be refracted as it enters and exits the dielectric bead. As a result, the ray will exit in a direction different from which it originated. Since light has a momentum associated with it, this change in direction indicates that its momentum has changed. Due to Newton's third law, there should be an equal and opposite momentum change on the particle. 1.25 µm silica spheres in DNA solutions



- (a) colloid particles in buffer solution(b) colloid particles + polymer
- (c) depletion interaction



Direct Observation of Aggregation in a Phase-Separating Colloid-Polymer Suspension

PMMA-latex in poly(styrene) solutions



FIG. 3. Phase diagram of the colloid-polymer suspension with the four regimes indicated.



Depletion Interaction in Liquid Crystal (Rod-like Particle) + Polymer System



Overlapping of depletion zone for sphere + rod fd virus (*d*=6.6 nm, *l*=880 nm) + PS latex (D=100 nm)





M. Adams, et al., Nature 393, 349 (1998).

Polymer chains confined into lamellar membrane slit



Polymer size in bulk state

$$R_{F3} \cong aN^{\nu} = aN^{3/5}$$
 generally, $\nu = \frac{3}{d_i + 2}$

This polymer chain is confined into lamellar slit

$$\begin{split} \overline{d} >> R_{\text{F3}} &\approx a N^{3/5} \\ \overline{d} < R_{\text{F3}} &\approx a N^{3/5} \\ \overline{d} < R_{\text{F3}} \\ R_{F2} &= R_{F3} f(\frac{R_{F3}}{\overline{d}}) = R_{F3} f(x) \end{split}$$

Scaling function f(x) has the following nature





Scaling law in polymer chain

A Gaussian chain is described by the segment length *b* and the number of segments *N*. We introduce a coarse grained parameter set. A new segment is composed of λ segments. In this case,

$$N \rightarrow \lambda^{-1} N \qquad b \rightarrow \lambda^{1/2} b$$

For *N*>>1, the macroscopic nature, such as mean square end-to-end distance, is invariant against this transformation.

$$f(\lambda^{-1}N,\lambda^{1/2}b) = f(N,b)$$

For excluded volume chain

$$f(\lambda^{-1}N,\lambda^{\nu}b) = f(N,b) \qquad \nu = 3/5$$

Generally,

 $A(\lambda^{-1}N,\lambda^{\nu}b) = \lambda^{x}A(N,b)$



Overlap concentration of polymer chains (dilute - semi-dilute boundary)

3 D case Overlap concentration ρ_3^* is expressed by $\rho_3^* \cong \frac{N}{R_{F3}^3} \cong \frac{N}{(aN^{3/5})^3} \cong a^{-3}N^{-4/5} \qquad \overline{\Phi}_{poly} = Na^3 / R_{F3}^3 = a^3 \rho_3$

$${}^{3}\overline{\Phi}_{poly}^{*} = N^{-4/5}$$

2 D case Overlap concentration ρ_2^* is expressed by

$$\rho_{2}^{*}(x) \approx \frac{N}{\overline{d}R_{F2}^{2}} = \frac{N}{\overline{d}(R_{F3}(\frac{R_{F3}}{\overline{d}})^{1/4})^{2}} = \frac{R_{F3}}{\overline{d}} \frac{N}{R_{F3}^{3}(\frac{R_{F3}}{\overline{d}})^{1/2}}$$
$$\approx \rho_{3}^{*} x^{1/2}$$
$$^{2}\overline{\Phi}_{poly}^{*} \approx a^{3}\rho_{2}^{*} = a^{3}(a^{-3}N^{-4/5})(R_{F3}/\overline{d})^{1/2} = a^{1/2}/(\overline{d}^{1/2}N^{1/2})$$
$$\frac{\overline{d}}{\overline{a}} \sim \frac{1}{N\overline{\Phi}_{poly}^{2}}$$

Semi-dilute solution of polymer chains

 $\xi_2 \cong (\frac{a^2}{\bar{d}}) \overline{\Phi}_{poly}^{-3/2}$

Blob model

3D

$$\begin{split} \Im \overline{\Phi}_{poly} &= R_{F3} \left(\frac{3\overline{\Phi}_{poly}^{*}}{\overline{\Phi}_{poly}}\right)^{m_{\xi 3}} & R_{F3} \cong a N^{3/5} \\ \Im \overline{\Phi}_{poly}^{*} = N^{-4/5} \\ \Im \overline{\Phi}_{poly}^{*} = N^{-4/5} \\ \text{Exponent } m_{\xi 3} \text{ should be independent of } N. \\ m_{\xi 3} &= 3/4 \\ & \xi_{3} \cong a \overline{\Phi}_{poly}^{-3/4} \\ & \overline{d} = \xi_{3} \cong a \overline{\Phi}_{poly}^{-3/4} & \overline{d}_{\overline{a}} \cong \overline{\Phi}_{poly}^{-3/4} \\ \text{2D} \\ & \Im (\overline{\Phi}_{poly}) = R_{F2} \left(\frac{2\overline{\Phi}_{poly}^{*}}{\overline{\Phi}_{poly}}\right)^{m_{\xi 2}} \\ & R_{F2} \cong a N^{3/4} \left(\frac{a}{\overline{d}}\right)^{1/4} \quad 2\overline{\Phi}_{poly}^{*} \cong a^{1/2} / (\overline{d}^{1/2} N^{1/2}) \end{split}$$

Polymer contribution to the free energy per unit bilayer area

1) 3DD regime $\overline{d} >> R_{F3}$ and $\overline{\Phi}_{poly} <<^{3} \overline{\Phi}_{poly}^{*}$ In this regime, the solution consists of a dilute solution of swollen polymer coil of radius R_{F3}

Depletion contribution

$$\overline{\Phi}_{poly}^{e\!f\!f} \cong \overline{\Phi}_{poly} \, \frac{\overline{d}}{\overline{d} - 2R_{F3}}$$

Perfect gas contribution corrected for the depletion effect

$$f_{3DD} \cong \frac{k_B T}{a^3} \overline{d} \frac{\overline{\Phi}_{poly}^{eff}}{N} \ln(\overline{\Phi}_{poly}^{eff})$$



2) 3D S-D regime $\overline{d} > a\overline{\Phi}_{poly}^{-3/4}$ and $\overline{\Phi}_{poly} > {}^{3}\overline{\Phi}_{poly}^{*}$

In this regime the chains overlap but still remain unsqueezed. The free energy consists of "bulk" semi-dilute polymer solution term and the depletion effect term (thickness of the depletion layer ~ ξ_3)

Bulk energy $\sim k_B T / \xi_3^3$

Depletion effect term

Osmotic pressure
$$\Pi \cong \frac{1}{\xi_3^3} k_B T$$

 $f_{3DSD}^{osm} = \Pi \xi_3 = \frac{k_B T}{\xi_3^2}$

$$f_{3DSD} = \beta \frac{k_B T}{a^3} \overline{\Phi}_{poly}^{9/4} \overline{d} + 2\gamma \frac{k_B T}{a^2} \overline{\Phi}_{poly}^{3/2}$$

3) 2DSD regime
$$a/(N\overline{\Phi}_{poly}^2) < \overline{d} < a\overline{\Phi}_{poly}^{-3/4}$$
 and $\overline{\Phi}_{poly} > {}^3\overline{\Phi}_{poly}^*$

In this regime, the chains are strongly compressed and the solution behaves as an effective two-dimensional semi-dilute solution.

Osmotic pressure contribution

$$f_{2DSD}^{osmo} \sim \frac{k_B T}{\xi_2^2}$$

Entropic confinement contribution

No. of blobs per pancake

$$\xi_2 \cong \overline{d}N_{pan}^{3/4} \implies N_{pan} \cong (\xi_2/\overline{d})^{4/3}$$

No. of pancake blob per unit area $1/\xi_2^2$

$$f_{2DSD}^{conf} \sim \frac{k_B T}{\xi_2^2} (\frac{\xi_2}{\bar{d}})^{4/3}$$

$$f_{2DSD} = \delta \frac{k_B T}{a^2} (\frac{a}{\overline{d}})^{2/3} \overline{\Phi}_{poly} + \eta \frac{k_B T}{a^2} (\frac{\overline{d}}{a})^2 \overline{\Phi}_{poly}^3 \qquad \delta \approx 2.22, \ \eta \approx 1.72$$



4) 2DD regime

 $\overline{d} < R_F$ and $\overline{\Phi}_{poly} < {}^3 \overline{\Phi}_{poly}^*$ or $\overline{d} << a/(N\overline{\Phi}_{poly}^2)$ and $\overline{\Phi}_{poly} >> {}^3 \overline{\Phi}_{poly}^*$ In this regime the polymer solution consists of a two-dimensional dilute gas of compressed polymer pancakes.

Entropic confinement contribution

No. of blobs per pancake

$$\xi_2 \cong \overline{d}N_{pan}^{3/4} \implies N_{pan} \cong (\xi_2/\overline{d})^{4/3}$$

No. of pancake blob per unit area $1/\xi_2^2$

$$f_{2DD}^{conf}\sim \frac{k_BT}{\xi_2^2}(\frac{\xi_2}{\overline{d}})^{4/3}$$

Two-dimensional dilute gas

$$f_{2DD}^{gas} = k_B T \frac{\overline{d}\overline{\Phi}_{poly}}{Na^3} \ln(\frac{\overline{d}\overline{\Phi}_{poly}}{Na})$$

$$f_{2DD} = \frac{k_B T}{a^2} \left[\frac{\overline{d} \overline{\Phi}_{poly}}{Na} \ln(\frac{\overline{d} \overline{\Phi}_{poly}}{Na}) + \alpha(\frac{\overline{d}}{\overline{d}})^{2/3} \overline{\Phi}_{poly} \right]$$

Layer compression modulus of a doped-solvent lamellar phase

 σ : surface fraction of guest component molecules per unit bilayer area

$$\tilde{f} = f - \mu \sigma$$

 μ : chemical potential of the guest component (conjugate to σ)

$$\bar{B}_{\mu} = d \frac{\partial^{2} \tilde{f}}{\partial \bar{d}^{2}} \qquad \bar{B}_{\mu}^{poly} = d \{ \frac{\partial^{2} f}{\partial \bar{d}^{2}} - \frac{\left[\frac{1}{\bar{d}} \frac{\partial f}{\partial \overline{\Phi}_{poly}} - \frac{\partial^{2} f}{\partial \overline{\Phi}_{poly} \partial \overline{d}}\right]^{2}}{\frac{\partial^{2} f}{\partial \overline{\Phi}_{poly}^{2}}} \}$$

1) 3DD regime

$$\overline{B}_{\mu}^{3DD} = -\frac{4k_B T R_F^2 d\overline{\Phi}_{poly}}{a^3 N \overline{d}^3} (1 + \ln[\frac{\overline{\Phi}_{poly}}{1 - 2(R_{F3}/\overline{d})}])^2$$
$$\approx -4 \frac{k_B T N^{1/5} d\overline{\Phi}_{poly}}{a \overline{d}^3}$$

2) 3DSD regime

$$\begin{split} \overline{B}^{3DSD}_{\mu} &= -d \frac{k_B T}{a \overline{d}^3} \frac{9 \rho^2 \overline{\Phi}_{poly}}{\frac{45}{16} \beta \overline{\Phi}^{1/4}_{poly} + \frac{3}{2} \rho \frac{a}{\overline{d}} \overline{\Phi}^{-1/2}_{poly}} \\ &\cong -\frac{16 \rho^2 k_B T}{5 \beta a} \frac{d}{\overline{d}^3} \overline{\Phi}^{3/4}_{poly} \end{split}$$

3) 2DSD regime

$$\begin{split} \overline{B}_{\mu}^{2DSD} &= d \left[\frac{25}{9} \frac{\delta k_B T}{a^4} (\frac{a}{\overline{d}})^{8/3} \overline{\Phi}_{poly} + \frac{\eta k_B T \overline{\Phi}_{poly}^3}{2a^4} - \frac{25 k_B T}{54a^4 \eta} (\frac{a}{\overline{d}})^{16/3} \overline{\Phi}_{poly}^{-1} \right] \\ &\cong - \frac{25 \delta^2 k_B T}{54\eta} \frac{d}{a^4} (\frac{a}{\overline{d}})^{16/3} \overline{\Phi}_{poly}^{-1} \end{split}$$

4) 2DD regime

$$\overline{B}_{\mu}^{2DD} = d \frac{5\alpha k_B T}{9\overline{d}^4} (\frac{a}{\overline{d}})^{1/3} \overline{\Phi}_{poly} (-5\alpha N + 8(\frac{\overline{d}}{a})^{5/3})$$
$$\approx -\frac{25\alpha^2 k_B T}{9} N \overline{\Phi}_{poly} (\frac{a}{\overline{d}})^{1/3} \frac{d}{\overline{d}^4}$$

In all four regimes, the polymer confinement contribution is negative.

Layer compression modulus contributed from other inter-membrane interactions

$$\overline{B}_{mem} = d \frac{\partial^2 V}{\partial \overline{d}^2}$$

van der Waals interaction

$$\overline{B}_{vdW} = -\frac{dk_B T}{12} \left(\frac{6}{(d-\delta)^4} - \frac{12}{d^4} + \frac{6}{(d+\delta)^4}\right)$$

Hydration interaction

$$\overline{B}_{hyd} = \frac{dH_0}{\lambda^2} e^{-\overline{d}/\lambda}$$

Electrostatic interaction

$$\overline{B}_{ele} = \frac{4k_B T d}{\pi \lambda_D^3 l_B} \gamma^2 e^{-\overline{d}/\lambda_D} \quad \text{if [NaCl]} \neq 0 \text{ and } \overline{d}/\lambda_D >> 1$$

Helfrich interaction

$$\overline{B}_{und} = \frac{3(k_B T)^2 d}{16\mu_{Hel}\kappa \overline{d}^4}$$

Total layer compression modulus

$$\overline{B} = \overline{B}_{mem} + \overline{B}_{\mu}^{poly}$$

Sample: CPCl(cetylpyridinium chloride)/hexanol/water + Polyvinylpyroridon



Scattering Profile Analysis

$$I(q) = 2\pi \frac{V}{d} \frac{P(q)S(q)}{q^2}$$

Structure Factor



Form factor

$$P_{neut}(q) = \frac{2\Delta\rho^2}{q^2} [1 - \cos(q\delta_{neut})e^{-q^2\sigma^2/2}]$$
$$P_{X-ray}(q) = \frac{4}{q^2} \{\Delta\rho_H[\sin[q(\delta_H + \delta_T)] - \sin(q\delta_T)] + \Delta\rho_T \sin(q\delta_T)\}^2$$



3DD - 3DSD



Electrostatic interaction - Helfrich interaction



Confinement of colloids in lamellar slits

Samples:

Surfactant : $C_{12}E_5$ $CH_3(CH_2)_{11}(OC_2H_4)_5OH$ At 5 wt% $\overline{d} \approx 80 \text{ nm}$

Colloidal Particle: Polystyrene Latex (a = 8 nm)

- Counter experiment

Polymer : Polyethylene Oxide (PEO) *M*w=20,000 *R*g~10 nm

Neutron Scattering Experiments Small Angle Neutron Scattering (SANS) Neutron Spin Echo (NSE)



Phase Diagram of $C_{12}E_5$ System

Neutron Scattering Facility at Tokai



Lamellar Membrane + Polymer System



Phase Diagram

SANS Profiles

Ligoure et al., J. Phys. II France 1997.



Membrane + Polymer System

Membrane + Colloids System

Estimation of Membrane-Colloid Interaction Potential: $f_{\rm mc}$

Layer compressibility: \overline{B}_{μ}

$$\overline{B}_{\mu} = d\left\{\frac{\partial^2 f_{mc}}{\partial \overline{d}^2} - \frac{\left[\frac{1}{\overline{d}}\frac{\partial f_{mc}}{\partial \overline{\Phi}} - \frac{\partial^2 f_{mc}}{\partial \overline{\Phi} \partial \overline{d}}\right]^2}{\frac{\partial^2 f_{mc}}{\partial \overline{\Phi}^2}}\right\}$$

Static Structure Factor

Dynamic Structure Factor

Estimation of Caillé parameter η

Caillé parameter η v.s. colloid concentration Φ_{c}

Estimation of bending modulus from NSE measurements

$$\frac{S(q,\tau)}{S(q,0)} = \exp[-(\Gamma\tau)^{2/3}]$$

$$\Gamma_q = 0.025\gamma_k \left(\frac{k_B T}{\kappa}\right)^{1/2} \frac{k_B T}{\eta_v} q^3$$

$$\gamma_k \approx 1 - 3\ln(q\xi)k_B T / 4\pi\kappa$$

$$\kappa / k_B T$$

$$C_{12}E_5 / \text{water} \sim 3.0$$

$$C_{12}E_5 / \text{water} + \text{colloidal particle} \sim 2.8$$

Bending modulus is independent of addition of colloidal particles.

Layer Compressibility

$$\eta = \frac{q_0^2 k_B T}{8\pi \sqrt{K\overline{B}}}$$

Model for Membrane-Colloid Interaction Potential: $f_{\rm mc}$

Suppression of Membrane Fluctuations by Confined Colloids

Effective volume fraction of confined colloids

$$\overline{\Phi}_c^{eff} = \overline{\Phi}_c \left(\frac{\overline{d}}{R_c}\right)^m$$

Restriction of membrane fluctuations

$$|u(r)| \leq \overline{d} \tanh^{n}(1/\alpha \overline{\Phi}_{c}^{eff})$$

$$\overline{\Phi}_{c}^{eff} << 1 \quad |u(r)| \leq \overline{d}$$

$$\begin{split} \overline{\Phi}_{c}^{eff} &\sim 1 \qquad | u(r) | \sim 0 \\ R(q) &= \frac{\langle | u_{q} |^{2} \rangle_{restricted}}{\langle | u_{q} |^{2} \rangle_{free}} \\ f_{mc} &= -\frac{1}{2} k_{B} T \ln R(q) = \frac{(k_{B} T)^{2}}{32\mu \kappa} \frac{1}{\overline{d}^{2} \tanh^{n}(1/\alpha \overline{\Phi}_{c}^{eff})} \\ \overline{B}_{mc} &= \frac{d}{32\mu} \frac{(k_{B} T)^{2}}{\kappa \overline{d}^{4}} \frac{g(\overline{\Phi}_{c}^{eff})}{h(\overline{\Phi}_{c}^{eff})} \\ \end{split}$$

Comparison between experiments and theory

 $\Phi_{\rm c}$

0.01% 0.02%

900 1000

800

Morphology transition induced by colloids

Polydisperse prolate core shell model

$$I(q) = N_m \int g(r) P_m(q,r) \, dr \, S_m'(q)$$

Form Factor

$$\left\langle P_m(q) \right\rangle = \int_0^1 \left| H(q,x) \right|^2 dx H(q,x) = \left(\rho_t - \rho_h\right) \frac{4}{3} \pi a_t R_t^3 \left[3 \frac{j(u_t)}{u_t} \right] + \left(b_h - b_s\right) \frac{4}{3} \pi a R^3 \left[3 \frac{j(u)}{u} \right] \qquad u_t = q R_t \sqrt{a_t^2 x^2 + (1 - x^2)} \qquad u = q R \sqrt{a^2 x^2 + (1 - x^2)}$$

